Structural probes to investigate the photomagnetic properties in cyanometalate compounds

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The rational design of molecular systems, which exhibit light-switchable physical properties is a subject of the intense research activity. Over the past, chemists have investigated the synthesis of magnetic and photoresponsive complexes through rational choices of cyanido-based building blocks. This approach has been extremely successful, and various molecular architectures have been obtained with remarkable properties such as spin crossover, ¹ electron-transfer process,² and photoinduced magnetism.^{3,4} To unambiguously elucidate the photo-induced mechanism, the use of structural X-ray probes is extremely helpful.

In this presentation, we will focus on two compounds exhibiting photomagnetic properties. The first system, a dinuclear pair, namely $[(Tp)Fe^{III}(CN)_3Co^{II}(PY5Me_2)]^+$, has been studied using a combination of X-ray Absorption Spectroscopy (XAS) and X-ray Magnetic Circular Dichroism (XMCD) measurements done at Fe and Co L_{2,3} edges and at K edges. These local structural measurements, as well the bulk magnetic measurements have shown the reproducibility and the full reversibility of the photo-induced and the thermal-induced electron transfer (figure).⁵ The second system is the mononuclear compound K₄[Mo^{IV}(CN)₈].2H₂O for which the photomagnetic properties can be only understood with the help of photocrystallography study that reveals important reorganization of the Mo coordination sphere.

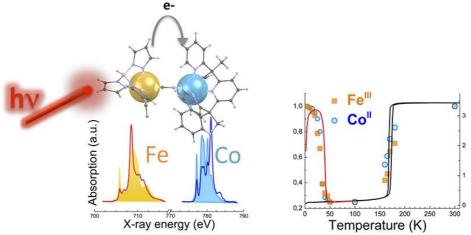


Figure. Left Top: Structure of $[(Tp)Fe(CN)_3Co(PY5Me_2)]^+$ and scheme of photo-induced electron transfer between Fe^{II} and Co^{III}. Left Bottom: L₃ edges XAS spectra measured after light excitation at 10 K (Fe^{III}: orange shape and Co^{III}: blue shape) and after thermal relaxation at 50 K (Fe^{III}: orange line and Co^{III}: blue line). Right: Comparison of Fe (yellow dots) and Co (blue dots) fractions obtained from XAS and magnetic properties under the variation of temperature (dark) and after light excitation red).

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